New Ring Opening Oxidation of Methyl Sterculate and Malvalate

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Air oxidation of cyclopropene fatty acid methyl esters, methyl sterculate and malvalate, resulted in the cleavage of cyclopropene ring, respectively, to afford a pair of conjugated enones containing a terminal methylene group.

In early studies on the seed germination inhibitors from Hibiscus, 1) we have isolated two new fatty acid methyl esters, methyl 10-octyl-9-oxo-10-undecenoate (1-a) and 9-octyl-8-oxo-9-decenoate (2-a), possessing a methylenic double bond conjugated with a carbonyl group. In continuation of our studies, we isolated two cyclopropene acid methyl esters, methyl sterculate (1)²⁾ and malvalate (2), 3) as the active constituents and found out that they were gently oxidized to a pair of conjugated enones with a methylenic double bond, respectively, by air. These results strongly supported our biogenetic aspect 1) on the production of 1-a and 2-a. We wish to report this new ring opening oxidation of the cyclopropenoids 1 and 2 in this communication.

$$\begin{array}{c} \text{CH}_2\\ \text{CH}_3(\text{CH}_2)_7\text{C} \longrightarrow \text{C}(\text{CH}_2)_7\text{CO}_2\text{CH}_3\\ \text{methyl sterculate (1)} \end{array}$$

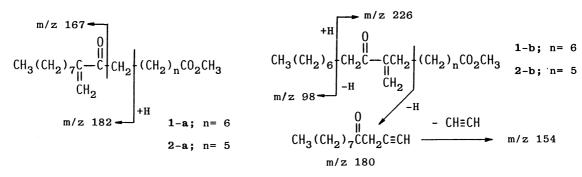
$$\text{methyl malvalate (2)}$$

Both the compounds 1 and 2 were sensitive to air or temperature, and therefore, by blowing air through their solutions or at 50 °C, they polymerized to give complex mixtures. But, without the bubbling air, they were slowly oxidized at room temperature to afford a pair of 1:1 mixture of conjugated enones 1-a and 1-b (50%), and 2-a and 2-b (48%), respectively, with a complex mixture after 3 days. After a week, they completely changed to give the enones in 71 and 68% yields, respectively.

The high-resolution mass spectra of 1-a and 1-b both indicated same

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molecular formula of $\mathrm{C}_{20}\mathrm{H}_{36}\mathrm{O}_3$ and their IR and UV spectra also resembled each other to show the presence of a conjugated enone group [ν_{max} 1670 and 1620 cm⁻¹ and λ_{max} 228 nm(ϵ 4800)], along with an ester carbonyl group (1730 cm⁻¹), which showed 1-a and 1-b to be isomers. Signals in the ¹H and ¹³C NMR spectra of **1-a** suggested the presence of a disubstituted methylenic double bond [δ 5.69(1H, br s) and 5.95(1H, s), and δ 149.1(s) and 123.3(t)] and two carbonyl groups [δ 202.5 and 176.9]. From these data, 1-a, having a conjugated enone system containing a terminal methylene group, was identified to be same compound as one isolated previously from Hibiscus. A comparison of the ^{1}H and ^{13}C NMR spectra of 1-b with those of 1-a indicated that the two compounds had nearly identical structures and differed from one another only in the location of the enone system.



Unequivocal support for their structures, 1-a and 1-b, were obtained from their EI mass spectra. The position of the CO group in 1-a has been assigned to C-9 as the significant α - and β -fission ions were obtained at m/z 167 and 182, respectively. On the other hand, 1-b showed a different fragmentation to obtain a characteristic rearranged peak at m/z 180 (41%) and a peak at m/z 154 (41%) due to a successive loss of CH≣CH, along with β -cleaved ion peaks to the CO group at m/z 226 (9%) and 98 (27%).

The remaining isomers, 2-a and 2-b, having a same molecular formula of $C_{1Q}H_{3d}O_3$, exhibited similar IR, UV, and NMR spectra to those of 1-a and 1-b, which proposed their structures, 2-a and 2-b. Their mass fragmentations also established them in a similar manner as above.

These two enone isomers would have been produced by an oxygen attack to either of the two olefinic carbons of cyclopropene rings in 1-a and 1-b, respectively.

References

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